

Auxiliary-field quantum Monte Carlo calculations of the molybdenum dimer

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Chemical accuracy is difficult to achieve for systems with transition metal atoms. Third row transition metal atoms are particularly challenging due to strong electron-electron correlation in localized d -orbitals. The Cr_2 molecule is an outstanding example, which we previously treated with highly accurate auxiliary-field quantum Monte Carlo (AFQMC) calculations [Purwanto *et al.*, *J. Chem. Phys.* 142, 064302 (2015)]. Somewhat surprisingly, computational description of the isoelectronic Mo_2 dimer has also, to date, been scattered and less than satisfactory. We present high-level theoretical benchmarks of the Mo_2 singlet ground state ($X^1\Sigma_g^+$) and first triplet excited state ($a^3\Sigma_u^+$), using the phaseless AFQMC calculations. Extrapolation to the complete basis set limit is performed. Excellent agreement with experimental spectroscopic constants is obtained. We also present a comparison of the correlation effects in Cr_2 and Mo_2 .

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I. INTRODUCTION

Transition metal (TM) dimers are of special interest theoretically and computationally. They fully exhibit the complexity of more complex TM materials (such as the formation of high-order bonds), but their relatively small sizes make them amenable to systematic and rigorous theoretical studies. Group VIB dimers are especially interesting, because the atom fragments are in the high-spin state (7S), and they form a closed shell ($^1\Sigma$) configuration in the molecular ground state. This results in a many-body spectrum with many nearly degenerate states, with strong electronic correlation effects. The Mo_2 molecule is similar to Cr_2 in that both are highly multiconfigurational in nature and require accurate treatment of both static and dynamic electron correlation. Because the $4d$ orbitals are not as localized as $3d$, the severity of electronic correlation is significantly reduced for Mo_2 . For example, a complete active space self-consistent field (CASSCF) treatment in Mo_2 recovers about 30% of the experimental binding energy, while with the same wave function, Cr_2 is not even bound. Nevertheless, the best quantum chemistry calculations for Mo_2 give widely varying predictions.¹⁻³

In this work, we present accurate theoretical calculations of Mo_2 potential energy curves (PECs) near the equilibrium geometry. We consider both the singlet ground state ($X^1\Sigma_g^+$) and triplet first excited state ($a^3\Sigma_u^+$), and calculate their spectroscopic properties. We employ the phaseless auxiliary-field quantum Monte Carlo (AFQMC) method⁴⁻⁷ in our calculations. Select benchmark calculations were also performed with exact free-projection (FP) AFQMC⁷⁻⁹ to help establish the accuracy of our calculations. A high-quality quadruple zeta (QZ) basis set is used for the majority of our calculations. Extrapolation to the complete basis set (CBS) limit is performed in combination with the triple zeta (TZ) basis results, following standard approaches. As shown below, the AFQMC results for the spectroscopic constants are in excellent agreement with experiment. We compare our results with those from other quantum chemistry approaches. The similarities and differences in electron-electron correlation between Mo_2 and Cr_2 are also examined.

The remainder of the paper proceeds as follows. The

methodology is discussed in Section II. Results of our Mo_2 calculations are presented in Section III. Section IV presents comparisons with previous many-body quantum chemistry results and an analysis of the relative sizes of the correlation energy contributions in Mo_2 and Cr_2 , as well as the effect of the trial wave function in the AFQMC calculations. We summarize our results in Section V.

II. METHODOLOGY

The AFQMC method projects the many-body ground state wave function from a given trial wave function.¹⁰⁻¹² It is implemented as random walks of Slater determinants, with orbitals expressed in a chosen single-particle basis. A phaseless approximation⁴ has been introduced to control the phase problem introduced by complex auxiliary-field sampling, resulting in a practical computational method that scales modestly with the system size [e.g. $\mathcal{O}(N^3)$ or $\mathcal{O}(N^4)$]. Its high accuracy has been demonstrated in many molecular and solid systems^{5,8,13-16} as well as model electronic systems.^{7,17}

As an orbitally based wave function method, the AFQMC theoretical framework has close relations to many-body quantum chemistry methods. When expressed in a one-particle gaussian type orbital (GTO) basis, both approaches use exactly the same Hamiltonian. Thus, many efficient techniques developed for correlated quantum chemistry methods can be directly imported. This was done, for example, by using Cholesky decomposition to remove a bottleneck in the handling of two-body interaction matrix elements for large basis sets¹⁸ and introducing frozen-core type of approaches to seamlessly embed the highly correlated AFQMC treatment within a larger mean-field environment.^{14,16}

Since static correlation is important in Mo_2 , it is desirable that the trial wave function account for this effect well. We perform CASSCF(12e,12o) calculations, which correlate 12 electrons in 12 active orbitals. The resulting multideterminant expansion is then truncated to retain about 93 – 95% of the total weight, yielding a multi-determinant wave function which is used without further optimization as our trial wave function. Typically this gives a trial wave function with 90 – 240 deter-

minants for the ground state, and 30 – 290 for the excited state. Since the number of determinants grows rapidly with bond stretching, we have to use a shorter cutoff (86 – 92%) at larger bond lengths (≥ 2.2 Å) in order to keep the number of determinants in a range that is easy to handle with the current state of our code.

We perform our AFQMC calculations using the all-electron atomic natural orbital relativistic correlation consistent (ANO RCC) GTO basis.¹⁹ Scalar relativity is treated with the Douglas-Kroll-Hess Hamiltonian. Spin-orbit effects on the dissociation energy were assessed with density functional theory calculations, using NWChem, with a cc-pwCVTZ-PP basis and Dolg’s energy-consistent spin-orbit relativistic effective core potential (ECP). There was essentially no effect on the dissociation energy (with the spin-orbit generalized gradient approximation PBE functional), so this was not pursued further. The Ar+3d¹⁰ atom-like core orbitals are frozen at the mean-field (CASSCF) level. The calculated results were obtained using basis sets up to the realistic QZ (8s7p5d3f2g1h contraction) basis,³ denoted ANO-QZ hereafter. As shown below, this basis provides excellent results for many quantities. The exception is the binding energy, which needs to be extrapolated to the CBS limit in order to make reliable comparisons with experiment.

III. RESULTS

We first present Mo₂ AFQMC results obtained with the ANO-QZ basis, before discussing the CBS-extrapolated results. Previously reported many-body calculations^{1–3} largely used multireference perturbative methods. The ANO-QZ CASPT2 (complete active space second-order perturbation theory) results of Borin *et al.*³ are among the most accurate. The exact and approximate AFQMC results, using the same basis, will provide a useful benchmark to these results.

Figure 1 shows the Mo₂ AFQMC binding-energy curves of the ground (singlet X) and excited (triplet a) states in the ANO-QZ basis. (The binding energy is defined as the difference between the molecular total energy and that of the two isolated atoms. It is shown as a function of $R_{\text{Mo-Mo}}$, the distance between the two Mo₂ nuclei.) AFQMC calculations of the Mo atoms were done using the ROHF trial wave function. For the ground state of the molecule, phaseless AFQMC calculations are performed using both UHF (AFQMC/UHF) and truncated CASSCF (AFQMC/CASSCF) trial wave functions. The computed binding energy curves are shown; also shown is the exact free projection AFQMC (FP-AFQMC) binding energy for a geometry near equilibrium. For the excited state, only AFQMC/CASSCF results are shown. Morse curves are fitted to these results and are shown as color bands whose width represents the combined stochastic and fitting uncertainties. Near the experimental equilibrium geometry, the ground state appears to exhibit stronger static correlation, as evidenced by the larger statistical error bar for the same amount of AFQMC computation. This is also consistent with the observation that the Hartree-Fock energy for the singlet state is higher than the triplet and that a larger number of de-

terminants are in the trial wave function for the ground state even though the same cutoff is applied when truncating both the ground- and excited-state CASSCF wave functions (see Sec. II).

While AFQMC/UHF overestimates the binding energy by ~ 0.25 eV compared to the exact FP-AFQMC, AFQMC/CASSCF shows excellent agreement. This establishes the high accuracy of the truncated CASSCF trial wave function for Mo₂. Similar behavior was found in our previous work on Cr₂,¹⁵ where a multi-determinant truncated CASSCF trial wave function was also required. In the more strongly correlated Cr₂ molecule, however, AFQMC/UHF overestimated the binding by ~ 0.9 eV near the equilibrium bond length, for a TZ basis.¹⁵ For comparison, the CASPT2 results³ in the same basis set are also shown in Fig. 1. We see that both the ground and excited states appear to be overbound, by 0.2 eV. In the ground state of Cr₂, CASPT2 results showed

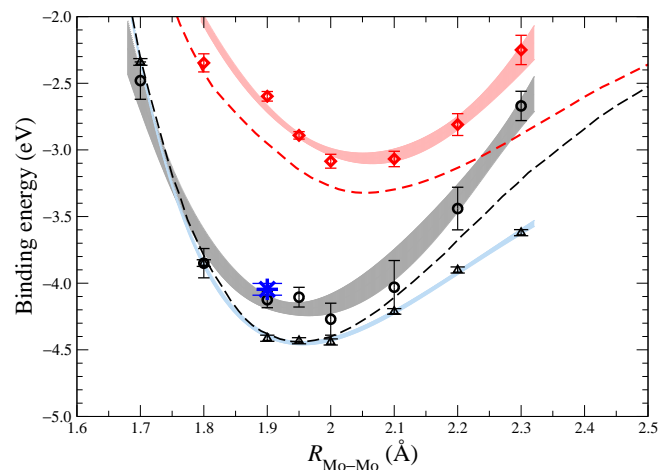


FIG. 1. AFQMC and CASPT2 Mo₂ PECs, both using the ANO-QZ basis, for the ground state (dark colors) and the lowest-energy triplet excited state (red). Shaded curves are Morse fits to the AFQMC results. CASPT2 PECs from Ref. 3 are shown as dashed lines. Ground-state AFQMC: AFQMC/UHF results are shown as triangle symbols with error bars and light blue shading; AFQMC/CASSCF are shown as black circles and grey shading. Only AFQMC/CASSCF results are shown for the excited state (red diamonds and red shading). Exact FP-AFQMC result for the ground state at $R(\text{Mo-Mo}) = 1.9$ Å is shown by the blue star symbol.

The CBS-extrapolated AFQMC/CASSCF PECs are shown in Fig. 2. As in our previous work,¹⁵ the QZ→CBS correction was obtained from AFQMC/UHF calculations using the ANO-TZ (7s6p4d2f1g) and ANO-QZ basis sets. We use a two-part scheme to extrapolate the many-body energies to the CBS limit:¹⁸ the exponential ansatz²¹ for the HF energies (with exponent $c = 1.63$) and the inverse cubic form²² for the correlation energies. The QZ→CBS correction increases the magnitude of the AFQMC binding energy by about 0.3 eV and 0.2 eV at the shortest and longest bond distances ($R_{\text{Mo-Mo}} = 1.7$ and 2.2 Å, respectively). In this geometry range, the correction is well approximated by a linear function of $R_{\text{Mo-Mo}}$. This correction was applied to ANO-QZ

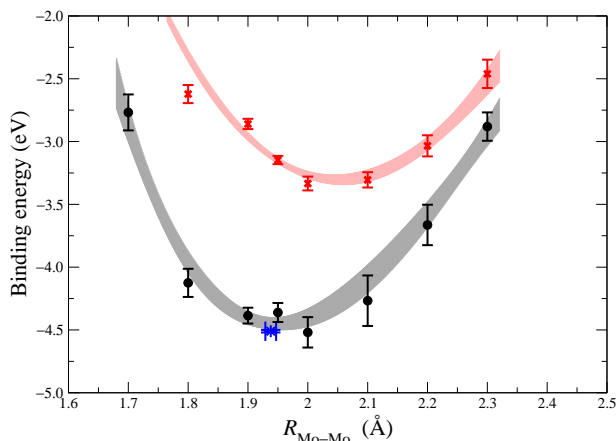


FIG. 2. CBS-extrapolated AFQMC/CASSCF PECs. Symbols, colors and shading are as in Fig. 1. The ground state experimental binding energy and bond length are given by the blue starred symbol, with the error bars representing the experimental uncertainties. The zero-point energy has been removed from the experimental binding energy.

AFQMC/CASSCF PECs to obtain the CBS limit. Test calculations showed that the excited state CBS correction was within error bars of the ground state CBS value, so we used the ground state correction for both.

Spectroscopic constants corresponding to Figs. 1 and Fig. 2 are given in Tables I and II for the ground and excited states, respectively. The tables also show results from experiment and from other high-level quantum chemistry many-body calculations. Our coupled cluster singles and doubles with perturbative triples [CCSD(T)] results were extrapolated to the CBS limit following the same procedure as described above, using CCSD(T) calculations for all basis sets; the CBS correction obtained this way was slightly smaller than, but consistent with, AFQMC/UHF CBS correction. For multi-reference perturbative calculations, the CBS extrapolation is less straightforward, since their correlation energies do not fit the inverse-cubic ansatz well, as discussed further below. Consequently, we made no attempt to apply the same CBS corrections to the perturbative results in Tables I and II; the values are listed in a separate column and correspond to the specified basis set.

As discussed in connection with Fig. 1, AFQMC/CASSCF is essentially exact near equilibrium. Thus, the AFQMC/CASSCF results in Tables I and II provide a benchmark for assessing the other quantum chemistry methods. At the CBS limit AFQMC/CASSCF is seen to be in excellent agreement with experiment. In contrast, other quantum chemistry results show considerable variance, especially for the molecular dissociation energy D_e . This is discussed further in the next section.

TABLE I. Ground state ($X^1\Sigma_g^+$) spectroscopic properties of Mo_2 computed using phaseless AFQMC and other quantum chemistry methods. D_e is the molecular dissociation energy, in units of eV (where the zero-point energy ~ 0.03 eV has been removed from the experimental value); R_0 is the equilibrium bond length (in Å); and ω_e is the harmonic vibrational frequency (in cm^{-1}). Unless otherwise indicated, the ANO-QZ basis (see text) is used. CBS extrapolation of D_e is shown also, when applicable.

Method	D_e	D_e (CBS)	R_0	ω_e
Multireference perturbation theory				
PT2-NEV (larger basis) ^{abc}	4.8845	—	1.9187	507.64
PT2-NEV ^{abd}	5.055	—	1.9198	506.09
PT3-NEV ^{ade}	3.9868	—	1.9500	461.54
PT2 ^{fg}	4.41	—	1.950	459
MRSDCI+ Q^h	3.9	—	1.993	447.5
CCSD(T)	3.85	4.06	1.913	549
AFQMC/UHF	4.45(1)	4.66(1)	1.955(4)	428(5)
AFQMC/CASSCF	4.20(5)	4.46(5)	1.95(2)	467(24)
Experiment		4.51(1) ⁱ	1.940(9) ^j	477.1 ^k

^a Ref. 2.

^b Strongly-contracted NEVPT2(12e,12o) method.

^c Basis set: full ANO basis (10s9p9d6f4g2h)

^d Basis set: ANO-QZ basis without the h functions (8s7p5d3f2g)

^e Strongly-contracted NEVPT3(12e,12o) method.

^f Ref. 3.

^g CASPT2(12e,12o) method.

^h Ref. 1. Calculations use an ECP with 5s5p4d1f basis (see cited article for detail).

ⁱ Ref. 23

^j Ref. 24

^k Ref. 25

TABLE II. Excited state ($a^3\Sigma_u^+$) spectroscopic properties of Mo_2 computed using phaseless AFQMC and other quantum chemistry methods. T_e is the excitation energy from the ground state (in units of eV); R_0 is the bond length at the PEC minimum (in Å); and ω_e is the harmonic vibrational frequency (in cm^{-1}). Unless otherwise indicated, the ANO-QZ basis (see text) is used. CBS extrapolation of T_e is shown also, when applicable.

Method	T_e	T_e (CBS)	R_0	ω_e
PT2 ^{ab}	1.105	—	2.063	393
AFQMC/CASSCF	1.15(6)	1.15(6)	2.05(1)	399(20)
Experiment		0.9947 ^c	—	393.7 ^c

^a Ref. 3.

^b CASPT2(12e,12o) method.

^c Ref. 26

IV. DISCUSSION

In this section, we analyze in more detail both the comparison of AFQMC with other methods and the comparison of the correlation effects in Mo_2 with Cr_2 . Correlation effects are greatly reduced in Mo_2 compared to Cr_2 . While CASSCF result is not bound for Cr_2 for an active space as large as (12e,28o), in Mo_2 CASSCF(12e,12o) already recovers 30% of the binding energy. For systems with strong static correlation,

multireference perturbation theory has often been the method of choice. As shown in Table I, however, the results depend rather sensitively on the perturbative implementation. Given that Mo_2 is considerably more benign than Cr_2 , it is perhaps somewhat surprising that it turns out to be rather challenging to the best quantum chemistry methods.

A. Comparison with other theoretical results

The CASPT2 calculation of Borin *et al.*,³ using the ANO-QZ basis and a CASSCF(12e,12o) active space zero-order wave function, overestimates D_e by ~ 0.2 eV. In Cr_2 , CASPT2(12,12) shows much larger overbinding of ~ 1.0 eV, based on direct comparisons with the exact FP-AFQMC.¹⁵ This trend has been attributed to the inadequacy of the active space chosen for the zeroth-order wave function.^{2,20} In Cr_2 , an improved zeroth-order wave function was obtained for a larger active space of 12 electrons and 28 orbitals, using a self-consistent density matrix renormalization group (DMRG) calculation.²⁰ The subsequent CASPT2 calculation²⁰ reduces, but does not eliminate, the discrepancy with exact FP-AFQMC, resulting in underbinding by ~ 0.4 eV.¹⁵ Multireference perturbative methods are also sensitive to the perturbative implementation. Results from an alternative perturbation treatment, using n -electron valence perturbation theory (NEVPT)²⁷ are also shown in Table I. The second-order PT2-NEV results show larger overbinding, while third-order PT3-NEV is underbound by ~ 0.2 eV. The D_e of PT2-NEV is reduced by 0.2 eV upon increasing the ANO basis size. This is opposite to AFQMC and CCSD(T), which show D_e increasing with basis size. The basis convergence of the perturbative calculations does not follow the empirical x^{-3} behavior, where x is the (correlation consistent) basis cardinal number. (Hence no CBS extrapolation is performed on the results, as mentioned earlier.) In view of these considerations, the agreement of the CASPT2 Mo_2 ANO-QZ calculation with the experimental binding energy is likely somewhat fortuitous.

Table I also presents results from multireference singles and doubles configuration interaction (MRSDCI) calculations,¹ with Q correction applied, and from CCSD(T). The MRSDCI+ Q used an ECP and a smaller basis. Single-reference CCSD(T), with restricted HF reference wave function, performs much better in Mo_2 than in Cr_2 . While CCSD(T) predicts that the Cr_2 molecule is unbound, for Mo_2 it yields respectable agreement with experiment. Nevertheless, the multireference character of the Mo_2 ground state is sufficiently strong that CCSD(T) still underestimates D_e by 0.45 eV, as shown in Table I.

For the Mo_2 ground state, all of the standard methods have difficulty obtaining an accurate dissociation energy. For the triplet excited state (Table II), we find that the AFQMC and CASPT2 results agree very well with each other and with experiment.

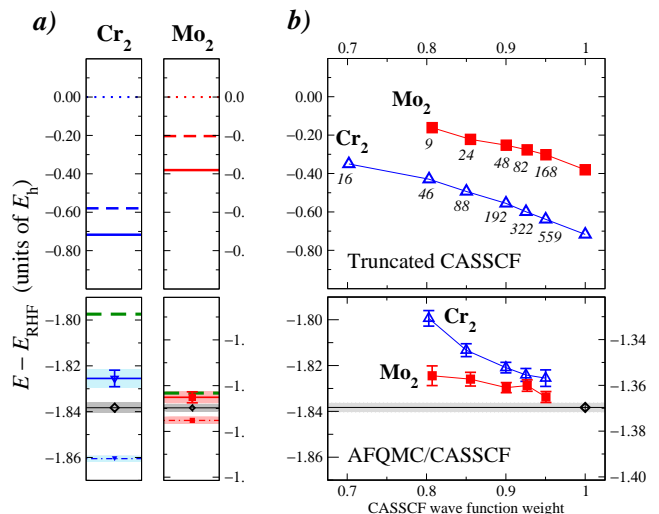


FIG. 3. The correlation energy in the Cr_2 and Mo_2 molecules. All energies are reported relative to the restricted HF (RHF) energy. Note that the energy offset in the lower panels of (a) and (b) aligns the (exact) FP-AFQMC correlation energies of Cr_2 and Mo_2 . a) Upper panels: RHF (dotted zero baseline), UHF (dashed line), and CASSCF (solid line) energies. Lower panels: CCSD(T) (dashed line), AFQMC/CASSCF at 95% weight cutoff (triangle and square), FP-AFQMC (diamond), AFQMC/UHF (circle); the AFQMC statistical uncertainties are indicated by the shading. b) Correlation energy as a function of the retained weight in the multi-determinant CASSCF wave function. Solid squares and open triangles denote the Mo_2 and Cr_2 results, respectively. Upper panel: variational energy of the truncated CASSCF wave function; numbers adjacent to the symbols give the number of determinants in the truncated wave function. Lower panel: the corresponding AFQMC/CASSCF correlation energies; statistical uncertainties indicated by the error bars. The exact FP-AFQMC energy from (a) is also shown. See the text for additional details.

B. Comparing Cr_2 and Mo_2

In this section, we quantitatively compare the effect of electron-electron correlation in Cr_2 and Mo_2 . We use the exact FP-AFQMC to benchmark the relative effects between the two molecules. Figure 3 compares the magnitude of electron-electron correlation effects in the Cr_2 and Mo_2 molecules. Results for Cr_2 were obtained using the cc-pwCVTZ-DK basis at the experimental bond length $R = 1.6788$ Å (see Ref. 15); the Cr_2 RHF energy is $-2098.533662 E_h$. Results for Mo_2 were obtained using the ANO-QZ basis near the experimental bond length $R = 1.9$ Å; the Mo_2 RHF energy is $-8091.069911 E_h$. Although the correlation energy is $\sim 0.5 E_h$ larger in Cr_2 than in Mo_2 , the Cr_2 UHF wave function recovers a larger fraction of the correlation energy, 32%, versus 15% in Mo_2 . The CASSCF wave function shows a similar but less pronounced trend, recovering 39% and 28% in Cr_2 and Mo_2 , respectively.

The stronger correlation effects in Cr_2 , however, are evident in the top panel of Fig. 3b: achieving 95% of the CASSCF total wave function weight requires retaining 559 determinants in Cr_2 but only 168 in Mo_2 . This is also evident in the larger CCSD(T) discrepancy in Cr_2 than in Mo_2 ,

$\sim 40 mE_h$ and $\sim 7 mE_h$, respectively. The dependence of the AFQMC/CASSCF energies on the quality of the trial wave function is shown in the bottom panel of Fig. 3b. The dependence is significantly stronger in Cr_2 , where at 95% cutoff, the total energy is still $\sim 13 mE_h$ higher than the exact value (this error is $\sim 5 mE_h$ for Mo_2). At the variational level, although the truncated wave function recovers more correlation energy in Cr_2 than in Mo_2 for the same weight cut, its performance is worse in AFQMC/CASSCF. We attribute this to the larger dynamic correlation energy that must be recovered in Cr_2 . Cancellation of errors between the molecule and atom AFQMC energies leads to better agreement in binding energy, however. For the best truncated CASSCF wave function, the error in the binding energy is $\sim 5 mE_h$ ($\sim 0.14 \text{ eV}$) for Cr_2 , and virtually exact for Mo_2 .

Basis set errors will modify the correlation energy recovered by the different methods. The mean-field HF energies are quite well converged for the basis sets used here. For the many-body calculations, we estimate (using AFQMC/UHF) the CBS shifts to be $\sim -100 mE_h$ and $\sim -70 mE_h$ for Cr_2 and Mo_2 , respectively. For the purpose of the above comparisons, however, the relative error between the various approximate and exact methods should not change significantly.

V. SUMMARY

We have presented an accurate calculation of the Mo_2 ground state ($X^1\Sigma_g^+$) and first triplet excited state ($a^3\Sigma_u^+$). We use the phaseless AFQMC method with the truncated

CASSCF trial wave function (AFQMC/CASSCF). Calculations were done using high-quality, realistic basis sets, and extrapolation to the CBS limit is performed. The resulting PECs and spectroscopic constants are in excellent agreement with experiment. Comparisons are made with other high-level quantum chemistry methods. We have also quantified the extent of strong electron correlations in both Cr_2 and Mo_2 molecules. Molybdenum is important in a variety of systems which can potentially exhibit exotic properties from strong correlation and topological effects. Our results can serve as a benchmark as theoretical and computational methods are developed and employed to treat such systems reliably.

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